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NEWS
     6 DEC 01 LISA now available on STN
NEWS
     7 DEC 09
                12 databases to be removed from STN on December 31, 2004
NEWS
    8 DEC 15 MEDLINE update schedule for December 2004
NEWS
NEWS 9 DEC 17
                ELCOM reloaded; updating to resume; current-awareness
                 alerts (SDIs) affected
    10 DEC 17
                COMPUAB reloaded; updating to resume; current-awareness
NEWS
                 alerts (SDIs) affected
    11 DEC 17
                SOLIDSTATE reloaded; updating to resume; current-awareness
NEWS
                 alerts (SDIs) affected
    12 DEC 17
                CERAB reloaded; updating to resume; current-awareness
NEWS
                 alerts (SDIs) affected
     13 DEC 17
                THREE NEW FIELDS ADDED TO IFIPAT/IFIUDB/IFICDB
NEWS
     14 DEC 30
NEWS
                EPFULL: New patent full text database to be available on STN
     15 DEC 30
                CAPLUS - PATENT COVERAGE EXPANDED
NEWS
                No connect-hour charges in EPFULL during January and
NEWS 16 JAN 03
                 February 2005
                 CA/CAPLUS - Expanded patent coverage to include Russia
NEWS
    17 JAN 11
                 (Federal Institute of Industrial Property)
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NEWS EXPRESS JANUARY 10 CURRENT WINDOWS VERSION IS V7.01a, CURRENT MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP), AND CURRENT DISCOVER FILE IS DATED 10 JANUARY 2005

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FULL ESTIMATED COST

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FILE COVERS 1907 - 24 Jan 2005 VOL 142 ISS 5 FILE LAST UPDATED: 23 Jan 2005 (20050123/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> reactor

389533 REACTOR

231356 REACTORS

L1 436322 REACTOR

(REACTOR OR REACTORS)

=> preheat?

L2 45882 PREHEAT?

=> 11(1)L2

L3 3537 L1(L)L2

=> ACRYLIC ACID

240719 ACRYLIC

1274 ACRYLICS

241085 ACRYLIC

(ACRYLIC OR ACRYLICS)

3927157 ACID

1461249 ACIDS

4401144 ACID

(ACID OR ACIDS)

L4 105288 ACRYLIC ACID

(ACRYLIC (W) ACID)

=> 13n and 14

9 L3N

L5 0 L3N AND L4

=> 13 and 14

L6 24 L3 AND L4

=> d 16 14-24 ti

L6 ANSWER 14 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN

TI Catalytic oxidation of olefins

L6 ANSWER 15 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN

TI Catalysts for preparation of unsaturated acids and aldehydes

ANSWER 16 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN 1.6 ΤI Unsaturated aldehydes and nitriles 1.6 ANSWER 17 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN TI Hydrocoumarin and coumarin flavorings and odorants ANSWER 18 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN L6 ΤI Catalyst for oxidation of olefins to unsaturated aldehydes or carboxylic acids ANSWER 19 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN L6 ΤI Oxidation of lower aliphatic aldehydes 1.6 ANSWER 20 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN ΤI Oxidation of propylene ANSWER 21 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN 1.6 TΤ Oxidation of α , β -unsaturated aldehydes in the presence of a catalyst containing vanadium, phosphorus, manganese, and oxygen ANSWER 22 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN L6Catalytic polymerization of acetylenes and polyenes TI ANSWER 23 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN L6 ΤI Acrylonitrile and acrylic acid from propylene ANSWER 24 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN L6 Methacrylic acid and its esters TΤ => nitrogen 563853 NITROGEN 3495 NITROGENS 1.7 566296 NITROGEN (NITROGEN OR NITROGENS) => 16 and 17 2 L6 AND L7 L8 => d 18 1-2 ti fbib abs ANSWER 1 OF 2 CAPLUS COPYRIGHT 2005 ACS on STN L8 TISafe and rapidly practicable cleaning procedure for catalysts and reactor materials used in gas-phase oxidation process in the synthesis of (meth) acrylic acid AN2003:548783 CAPLUS DN 139:101561 Safe and rapidly practicable cleaning procedure for catalysts and reactor TImaterials used in gas-phase oxidation process in the synthesis of (meth) acrylic acid Hammon, Ulrich; Schroeder, Juergen IN BASF AG, Germany PA so Ger. Offen., 4 pp. CODEN: GWXXBX DT Patent LA German FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE _____ ----_____ -----PΙ DE 10217325 A1 20030717 DE 2002-10217325 20020418 DE 2002-10217325 A safe and rapidly practicable procedure for a renewed preheating

and/or after-cooling of a reaction zone used in a catalytic gas-phase oxidation process used in the synthesis of (meth)acrolein and/or (meth)

acrylic acid, especially avoiding a damage of catalyst and reactor material, may be carried out, if the oxygen content of the gas stream used amts. maximally 9 weight%, preferably ≤5 weight%, whereby the gas stream may be the circulating gas, optionally mixed with air.

ANSWER 2 OF 2 CAPLUS COPYRIGHT 2005 ACS on STN L8 ΤI Acrylonitrile and acrylic acid from propylene AN 1964:425012 CAPLUS DN 61:25012 OREF 61:4228h,4229a Acrylonitrile and acrylic acid from propylene Foster, Robert E. E. I. du Pont de Nemours & Co. PΑ SO 4 pp. Patent DTUnavailable LA PATENT NO. KIND DATE APPLICATION NO. ---------_____ 19640121 US PΙ US 3118927 The problem of temperature control and localized concentration of gaseous reactants in exothermic, fixed-bed catalytic gas-phase reactions for producing CH2:CHCN

(I) and CH2:CHCO2H (II) was avoided by diffusing one of the gases through a porous diffuser directly into the catalyst bed and contacting the diffused gas with the other reactant in the bed. Thus, a reactor (shown) was charged with 100 ml. of a K-AgCN-Ca(OH)2 on low-Fe silica (8-14 mesh) catalyst (U.S. 3,023,226, CA 57, 14948g) and 100 ml. acid-washed 8-14 mesh calcined quartz as a preheater. The catalyst was activated by heating at 400-20° (argon atmospheric). A mixture of argon (11 ml./sec.) and CH2:CHMe (III) (3.5 ml./sec.) was passed through the bed, and NO (2.7 ml./sec.) passed through a porous glass diffuser buried within the bed kept at 425-50°. The conversion of NO into I was 11% at 80% yield, the space time being 7 g./l. catalyst/hr. Several modifications were also described.

=> preheated gas 23047 PREHEATED 1405021 GAS 481598 GASES 1577505 GAS (GAS OR GASES) L9 268 PREHEATED GAS (PREHEATED (W) GAS) => 16 and 19 L10 0 L6 AND L9 => 14 and 19 L11 0 L4 AND L9 => preheated reactor 23047 PREHEATED 389533 REACTOR 231356 REACTORS 436322 REACTOR (REACTOR OR REACTORS) 18 PREHEATED REACTOR T.12 (PREHEATED (W) REACTOR)

0 L12 AND L4

=> logoff hold COST IN U.S. DOLLARS

=> 112 and 14

L13

SINCE FILE

TOTAL

19591029

FULL ESTIMATED COST 35.82 SESSION

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL
ENTRY SESSION
CA SUBSCRIBER PRICE -1.46 -1.46

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STN INTERNATIONAL SESSION SUSPENDED AT 07:45:57 ON 24 JAN 2005

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COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	35.82	36.03
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE ENTRY	TOTAL SESSION
CA SUBSCRIBER PRICE	-1.46	-1.46

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(FILE 'HOME' ENTERED AT 07:35:05 ON 24 JAN 2005)

FILE 'CAPLUS' ENTERED AT 07:35:17 ON 24 JAN 2005

436322 REACTOR T.1 L2 45882 PREHEAT? 3537 L1(L)L2 L3 105288 ACRYLIC ACID L40 L3N AND L4 L5 24 L3 AND L4 L6 566296 NITROGEN L7L8 2 L6 AND L7 268 PREHEATED GAS L9 0 L6 AND L9 L100 L4 AND L9 L1118 PREHEATED REACTOR L12 0 L12 AND L4 L13

=> d 16 1-14 ti

- L6 ANSWER 1 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Process for purification of waste gases containing oxygen and combustible components by thermal oxidation
- L6 ANSWER 2 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Process and apparatus for emulsion polymerization
- L6 ANSWER 3 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Safe and rapidly practicable cleaning procedure for catalysts and reactor materials used in gas-phase oxidation process in the synthesis of (meth) acrylic acid

- L6 ANSWER 4 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Longitudinal-flow reactor for oxidation reactions consisting of tube bundles designed for heat exchange
- L6 ANSWER 5 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Development of an ultra-rapid reactor for superabsorbent polymer
- L6 ANSWER 6 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Alternating copolymers with good light transmittance and their preparation
- L6 ANSWER 7 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Accelerated preparation of acrylate esters
- L6 ANSWER 8 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Pressure-sensitive emulsion adhesive comprising a polymer system having good tack and peel strengths and enhanced shear strength
- L6 ANSWER 9 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Apparatus and method for producing acrylic acid
- L6 ANSWER 10 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Cycloaliphatic acrylate compositions
- L6 ANSWER 11 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Continuous manufacture of acrylic polymer foam-based insulating material
- L6 ANSWER 12 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Catalytic production of acrylic or methacrylic acid
- L6 ANSWER 13 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Oxidation of propylene to acrolein and acrylic acid
- L6 ANSWER 14 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Catalytic oxidation of olefins

=> d 16 9 ti fbib abs

- L6 ANSWER 9 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Apparatus and method for producing acrylic acid
- AN 1983:471293 CAPLUS
- DN 99:71293
- TI Apparatus and method for producing acrylic acid

TETATO

- IN Dutkay, Ervin
- PA Intreprinderea Chimica, Risnov, Rom.
- SO Rom., 3 pp. CODEN: RUXXA3

DAMENIM NO

- DT Patent
- LA Romanian
- FAN.CNT 1

with

PI RO 76048 B 19810430 RO 1979-96377		19790125
RO 1979-96377	Α	19790125

AB Acrylic acid [79-10-7] is manufactured by continuously passing acrylamide sulfate [18185-97-2] containing 1.5-2.5 mol H2SO4 and polymerization inhibitor through a packed column, while water (preheated to the reaction temperature) is passed countercurrently through the column,

the residence time in the column being 0.9-1.5 h, the column temperature being 90-120°, and the column pressure being 0.85-0.97 mm, followed by allowing the reaction mixture to drain into a reactor from the column, with resonance time in reactor being 2-4 h, the reactor temperature being 140-160°, and the reactor

A DDT TOAMTON NO

pressure being 0.85-0.97 mm. This single-step process gives high yields (e.g., 92.8%) and can be automated.

=> d 16 7 ti fbib abs

- L6 ANSWER 7 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Accelerated preparation of acrylate esters
- AN 1990:78163 CAPLUS
- DN 112:78163
- TI Accelerated preparation of acrylate esters
- IN Powanda, Thomas M.; Imes, Robert H.; Collins, George L.
- PA Hoechst Celanese Corp., USA
- SO Eur. Pat. Appl., 7 pp.
- CODEN: EPXXDW
- DT Patent
- LA English
- FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	_	DATE
PI		A3	19910605	EP 1988-311335	_	19881130
	R. 52, 52, 1R,	05, 11	, 112	US 1988-168564	Δ	19880307
	US 4859792	A	19890822			
	00 1003/71			US 1986-891991		
	CA 1326684	A1	19940201			
				US 1988-168564	Α	19880307
	JP 01299252	A2	19891204	JP 1989-54844		19890307
				US 1988-168564	Α	19880307
				US 1988-168565	Α	19880307
PATE	NT FAMILY INFORMATIO	N:				
FAN						
				APPLICATION NO.		DATE
ΡI	EP 331844			EP 1988-311320		19881130
	R: BE, DE, FR,		· -			
	,,	,	•	US 1988-168565	Α	19880307
	US 4868329	A	19890919	US 1988-168565		19880307
				US 1986-891990	A1	19860801
	CA 1326683	A1	19940201	CA 1988-584525		19881130
				US 1988-168565	Α	19880307
	JP 01299252	A2	19891204	JP 1989-54844		19890307
				US 1988-168564	Α	19880307

AB The title esterification of an aliphatic hydroxy compound with acrylic acid (I) comprises adding one reactant to a reactor and heating from 100° to b.p. of I, adding the other reactant over time during which ≥65% H2O of reaction is removed, and then heating to completion of the reaction. Thus, 1 mol trimethylolpropane was preheated to 135-150° and added over 110 min to 3.3 mol I heated to 120° and H2O distillate was removed after 50 min, and the reaction continued for 5 h until the acrylate had acid number 0.11, free I content 0.013%, and viscosity 135 cPs. A process for preparation of the above acrylate not using preheating required 10 h to complete the reaction.

US 1988-168565

A 19880307

=> d 16 12-14 ti fbib abs

- L6 ANSWER 12 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Catalytic production of acrylic or methacrylic acid
- AN 1972:46658 CAPLUS
- DN 76:46658
- TI Catalytic production of acrylic or methacrylic acid

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PA Societa Italiana Resine S.p.A.
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SO Brit., 5 pp. CODEN: BRXXAA

DT Patent LA English

FAN.CNT 1

Acrylic acid (I) [79-10-7] and methacrylic acid (II) [79-41-4] were prepared in improved yields by treating poly(oxymethylene) diacetate (III) and poly(oxymethylene) dipropionate (IV), resp., in the presence of an inert gas in the vapor phase with aluminum silicate (V). AcOH and III were introduced at 370-90.deg. to a reactor containing V, preheated at 300-450.deg. to give 75% yield of I. A mixture of IV and EtCO2H was introduced at 390.deg. with N to a reactor containing V, pretreated with LiOH, to give 90% yield of II.

L6 ANSWER 13 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN

TI Oxidation of propylene to acrolein and acrylic acid

AN 1971:88364 CAPLUS

DN 74:88364

TI Oxidation of propylene to acrolein and acrylic acid

IN Eden, Jamal S.

PA Goodrich, B. F., Co.

SO Ger. Offen., 8 pp. CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

L MI	N.CIVI I				
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	DE 2031618	A	19710121	DE 1970-2031618	19700626
				US 1969-838349	A 19690701
	US 3634502	A	19720111	US 1969-838349	19690701
				1	A
	NL 7009746	A	19710105	NL 1970-9746	19700701
				US 1969-838349	A 19690701
	GB 1275186	Α	19720524	GB 1970-1275186	19700701
				US 1969-838349	A 19690701

Acrolein and CH2:CHCO2H were prepared by oxidation of propylene with oxy gen in the presence of Mo oxide, TeO2, and BPO4 at 375-415°. Thus, 123 g B(OH)3 was mixed with 85% H3PO4 in H2O, the mixture eva pd., and calcined 16 hr at 400° to give BPO4, which was mixed with a su supension of 203.29 g H2MoO4 in H2O and 63.84 g TeO2. The mixture was evaporated and calcined 16 hr at 400° and 21 hr at 450° to give a catalyst containing Mo oxide 75, TeO2 25, and BPO4 50 moles. A reactor was filled with 80 ml catalyst; steam 3.96, propylene 1, and oxy gen 4.02 moles (all preheated at 200-50°) were introduced, and the mixture w as heated 49 sec at 400° to give 49.82% acrolein and 31.80% CH2:CHC O2H at 100% propylene conversion.

L6 ANSWER 14 OF 24 CAPLUS COPYRIGHT 2005 ACS on STN

TI Catalytic oxidation of olefins

AN 1970:100057 CAPLUS

DN 72:100057

TI Catalytic oxidation of olefins

IN Trapasso, Louis E.; Wenrick, John D.

PA Goodrich, B. F., Co.

SO U.S., 2 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

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KIND
                                DATE
                                          APPLICATION NO.
                                                                   DATE
     PATENT NO.
     _____
                        - - - -
                               ------
                                           ______
                                                                   -----
                               19700224 US 1966-561043 19660628
US 1966-561043 A 19660628
PΙ
     US 3497553
     Propylene and isobutylene were oxidized in the presence of a catalyst
AB
     having molar ratios of NiO 0.5-5.0, Cr2O3 0.5-2.0, and TeO2 and Mo2O3
     0.5-4.0, and a Ni:Mo mole ratio of 0.1-1.0:1. Thus, Ni(NO3)2.6H2O 218.1,
     Cr(NO3)3.9H2O 150.1, and (NH4)6Mo7O24.4H2O 330 g were dissolved in 150,
     100, and 440 ml distilled H2O, resp. The molybdate solution was added slowly
to
     a mixture of the Ni and Cr solns., and 29.9 g powdered TeO2 was added. The
     paste obtained was dried at 100° in vacuo, baked for 16 hr at
     400°, and ground to give a catalyst with a 2:1:0.5-5.0 Ni:Cr:Te:Mo
     atomic ratio. A supported catalyst was prepared by adding 375 g aqueous
colloidal
     dispersion of microspheroidal silica containing 30-5% SiO2 (Ludox H.S).
     Propylene was oxidized in a fluidized-bed reactor containing 60 ml
     catalyst. Steam (0.127 mole/hr H2O) at 200-50° was passed into the
     reactor, and 0.0795 mole/hr 0 and 0.0317 mole/hr propylene were
     added, preheated to 200-50°, and passed into the
     reactor (Ni-Cr-Te-Mo-Si atomic ratio, temperature, % propylene conversion,
     % yield acrolein, and % yield acrylic acid given):
     1:1:1:4:2.5, 425°, 94.9, 61.48, 22.97: 1:1:2:4:2.5, 430°,
     66.47, 83.24, 10.41; 4:2:1:5:0, 385°, 83.72, 35.51, 33.31;
     4:1:1:6:0, 395°, 89.76, 39.85, 34.89. The contact time was 18 sec.
=> d his
     (FILE 'HOME' ENTERED AT 07:35:05 ON 24 JAN 2005)
     FILE 'CAPLUS' ENTERED AT 07:35:17 ON 24 JAN 2005
L1
         436322 REACTOR
L2
         45882 PREHEAT?
L3
          3537 L1(L)L2
        105288 ACRYLIC ACID
L4
L5
             0 L3N AND L4
L6
             24 L3 AND L4
L7
        566296 NITROGEN
L8
             2 L6 AND L7
L9
           268 PREHEATED GAS
L10
             0 L6 AND L9
L11
             0 L4 AND L9
L12
            18 PREHEATED REACTOR
L13
             0 L12 AND L4
=> \gas
       1405021 GAS
       481598 GASES
       1577505 \GAS
L14
                 (GAS OR GASES)
=> gas
       1405021 GAS
       481598 GASES
       1577505 GAS
L15
                 (GAS OR GASES)
=> 13(1)115
         1949 L3(L)L15
L16
=> /2(1)13
MISSING TERM FOR FIELD QUALIFICATION BEFORE '/2'
The search profile entered contains a field qualifier, e.g., '/AU',
with no term preceding it.
```

=> 12(1)13

L17 3537 L2(L)L3

=> 12(115)

MISSING OPERATOR 'L2 (L15'

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

=> 12(1)115

L18 15113 L2(L)L15

=> 117 and 118

L19 1953 L17 AND L18

=> logoff hold

COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION FULL ESTIMATED COST 76.48 76.69 SINCE FILE DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) TOTAL ENTRY SESSION CA SUBSCRIBER PRICE -5.11 -5.11

SESSION WILL BE HELD FOR 60 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 08:09:36 ON 24 JAN 2005

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COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	76.48	76.69
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	
CA SUBSCRIBER PRICE	-5.11	-5.11
=> logoff hold		
COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	76.48	76.69
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-5.11	-5.11

SESSION WILL BE HELD FOR 60 MINUTES STN INTERNATIONAL SESSION SUSPENDED AT 08:39:05 ON 24 JAN 2005

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Welcome to STN International! Enter x:x LOGINID:SSSPTA1623PAZ PASSWORD: * * * * * RECONNECTED TO STN INTERNATIONAL * * * * * SESSION RESUMED IN FILE 'CAPLUS' AT 09:11:34 ON 24 JAN 2005 FILE 'CAPLUS' ENTERED AT 09:11:34 ON 24 JAN 2005 COPYRIGHT (C) 2005 AMERICAN CHEMICAL SOCIETY (ACS) COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION FULL ESTIMATED COST 76.48 76.69 DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL ENTRY SESSION CA SUBSCRIBER PRICE -5.11 -5.11 => d his (FILE 'HOME' ENTERED AT 07:35:05 ON 24 JAN 2005) FILE 'CAPLUS' ENTERED AT 07:35:17 ON 24 JAN 2005 L1 436322 REACTOR L245882 PREHEAT? L3 3537 L1(L)L2 105288 ACRYLIC ACID L4L5 0 L3N AND L4 L6 24 L3 AND L4 566296 NITROGEN L7 L8 2 L6 AND L7 268 PREHEATED GAS L9 L10 0 L6 AND L9 L110 L4 AND L9 18 PREHEATED REACTOR L12L13 0 L12 AND L4 1577505 \GAS L141577505 GAS L15 L16 1949 L3(L)L15 L17 3537 L2(L)L3 L18 15113 L2(L)L15 L19 1953 L17 AND L18 => startup 5235 STARTUP 129 STARTUPS L20 5314 STARTUP (STARTUP OR STARTUPS) => **11(1)120** 1467 L1(L)L20 L21 => 12(1)121 L22 23 L2(L)L21 => d 122 10-23 ti L22 ANSWER 10 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN Process for starting up coal gasification reactors L22 ANSWER 11 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN Design construction and operation of a full scale downflow fixed film reactor using hogwaste substrate

- L22 ANSWER 12 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Plasma engineering analysis of the Tennessee tokamak
- L22 ANSWER 13 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Plasma engineering studies for Tennessee Tokamak (TENTOK) fusion power reactor
- L22 ANSWER 14 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Design of a reactor system for the synthesis of titanium diboride
- L22 ANSWER 15 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Sodium fill of FFTF
- L22 ANSWER 16 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Methanization reactor
- L22 ANSWER 17 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Kinetic study of the catalytic decomposition of liquid hydrazine. Effect of the initial temperatures of the catalytic bed and hydrazine for different ignition pressures
- L22 ANSWER 18 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Wet air oxidation system for strong sludges and liquors
- L22 ANSWER 19 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Analysis of filling accidents in Molten Salt Reactor Experiment
- L22 ANSWER 20 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Release of fission products from reactor fuels during transient accidents simulated in TREAT
- L22 ANSWER 21 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Operating problems in the production of petroleum naphthalenes from catalytic gas oil
- L22 ANSWER 22 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Fungicidal tribasic copper chloride
- L22 ANSWER 23 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI The disposal of radioactive waste
- => 14 and 122
- L23 0 L4 AND L22
- => d 122 10 ti fbib abs
- L22 ANSWER 10 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Process for starting up coal gasification reactors
- AN 1991:210473 CAPLUS
- DN 114:210473
- TI Process for starting up coal gasification reactors
- IN Duerlich, Manfred; Enders, Heinz; Wehner, Olaf; Findeisen, Hartmut; Thieme, Gerd; Toufar, Walter; Scholz, Guenter; Sowka, Karl; Graf, Hermann; et al.
- PA VEB Gaskombinat Schwarze Pumpe, Germany
- SO Ger. Offen., 5 pp.
 - CODEN: GWXXBX
- DT Patent
- LA German
- FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	DE 4013739	A1	19910131	DE 1990-4013739	19900428

			DD 1989-331077	A	19890724
DD 285989	A5	19910110	DD 1989-331077		19890724
DD 285989	B5	19940414			
CZ 280247	B6	19951213	CZ 1990-3597		19900719
			DD 1989-331077	Α	19890724
CN 1049024	Α	19910206	CN 1990-104821		19900724
			DD 1989-331077	Α	19890724

AB The coal gasification reactor start-up fuel mixture composition is varied during startup to steadily increase reactor temperature without forming an explosive mixture The mixture may be preheated or heated using steam. Inert gases, e.g., N2 or CO2, are used to increase temperature and prevent problems.

=> d 122 1-9 ti

- L22 ANSWER 1 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Liquid water and air injection for improved management of an autothermal reformer
- L22 ANSWER 2 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Fuel cell power plant and its startup method
- L22 ANSWER 3 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Steady-State and Dynamic Effects of Design Alternatives in Heat-Exchanger/Furnace/Reactor Processes
- L22 ANSWER 4 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Radially corrugated catalyst unit with rotatable start-up burner
- L22 ANSWER 5 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Control of an autothermal network of nonstationary catalytic reactors
- L22 ANSWER 6 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Development of an advanced startup procedure for a PIUS-type reactor
- L22 ANSWER 7 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Catalytic air purifiers
- L22 ANSWER 8 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI The simulation test to start up the PIUS-type reactor from isothermal fluid condition
- L22 ANSWER 9 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Limiting cases and approximate solutions for fixed-bed reactors with periodic flow reversal

=> d 122 6 ti fbib abs

- L22 ANSWER 6 OF 23 CAPLUS COPYRIGHT 2005 ACS on STN
- TI Development of an advanced startup procedure for a PIUS-type reactor
- AN 1998:643318 CAPLUS
- DN 129:322394
- TI Development of an advanced startup procedure for a PIUS-type reactor
- AU Ito, Takahiro; Oyamatsu, Kazuhiro; Tsuji, Yoshiyuki; Tamaki, Masayoshi; Kukita, Yutaka
- CS Dep. Nuclear Eng., Nagoya Univ., Nagoya, 464-8603, Japan
- SO Journal of Nuclear Science and Technology (1998), 35(8), 554-563 CODEN: JNSTAX; ISSN: 0022-3131
- PB Atomic Energy Society of Japan
- DT Journal
- LA English
- AB An advanced **startup** procedure for the PIUS-type **reactor** has been developed. The procedure is related to the way to isolate the

primary loops from the borated **reactor** pool by establishing stable hot/cold water interfaces in the so-called d. lock sections. The procedure starts with accumulating **preheated** water in the high points of the steam-generator-side legs. Then, by restarting the **reactor** coolant pumps, the primary loops can be isolated from the pool as the primary loops reaches a uniformly higher temperature than the pool water. The addnl. components required for this procedure are only a low-pressure grade heater and a pump of small capacities. Since the isolation is achieved with the d. locks left open, the core shutdown and cooling capabilities by means of the natural circulation of borated water are maintained in case of any abnormal events during **startup**. The feasibility and the predictability of this procedure were investigated by running an experiment in a scaled single-loop facility and conducting an anal. using a 1-dimensional model. Both in the experiment and in the anal., the primary loop was successfully isolated from the pool.

RE.CNT 13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> logoff hold		
COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	94.18	94.39
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-6.57	-6.57

SESSION WILL BE HELD FOR 60 MINUTES STN INTERNATIONAL SESSION SUSPENDED AT 09:15:41 ON 24 JAN 2005

Connecting via Winsock to STN

Welcome to STN International! Enter x:x

LOGINID:SSSPTA1623PAZ

PASSWORD:

* * * * * RECONNECTED TO STN INTERNATIONAL * * * * * * SESSION RESUMED IN FILE 'CAPLUS' AT 09:39:38 ON 24 JAN 2005 FILE 'CAPLUS' ENTERED AT 09:39:38 ON 24 JAN 2005 COPYRIGHT (C) 2005 AMERICAN CHEMICAL SOCIETY (ACS)

COST IN U.S. DOLLARS FULL ESTIMATED COST	SINCE FILE ENTRY 94.18	TOTAL SESSION 94.39
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=> logoff hold COST IN U.S. DOLLARS FULL ESTIMATED COST	SINCE FILE ENTRY 94.18	TOTAL SESSION 94.39
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SESSION WILL BE HELD FOR 60 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 09:39:50 ON 24 JAN 2005

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L2	15944	shell near3 tube	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L3	1771	L1 and L2	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L4	472	L1 same L2	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L5	12	"1089353"	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L6	550	562/598.ccls.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L7	0	L6 and L4	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L8		L6 and L3	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L9	359007	reactor	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L10	12311	L1 same L9	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L11	4	L6 and L10	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
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L13	0	("6651731").URPN.	USPAT	OR	ON	2005/01/24 13:01
L14	1	"3118927".pn.	USPAT	OR	ON	2005/01/24 13:01

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L16	4	"3634502".pn.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L17	2	"6676808".pn.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
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L20	1	"4341600".PN.	USPAT; USOCR	OR	ON	2005/01/24 13:01
L21	1	"4365081".PN.	USPAT; USOCR	OR	ON	2005/01/24 13:01
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L23	1	"4369097".PN.	USPAT; USOCR	OR	ON	2005/01/24 13:01
L24	1	"4986884".PN.	USPAT; USOCR	OR	ON	2005/01/24 13:01
L25	1	"6348135".PN.	USPAT; USOCR	OR	ON	2005/01/24 13:01
L26	1	"6348135".PN.	USPAT; USOCR	OR	ON	2005/01/24 13:01
L27	3	"3762465".pn.	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
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L29	38395	startup	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L30	1133	L9 same L29	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01
L31	87	L1 same L30	US-PGPUB; USPAT; EPO; JPO; DERWENT	OR	ON	2005/01/24 13:01

L32	0	L6 and L31	US-PGPUB;	OR	ON	2005/01/24 13:01
			USPAT;			
			EPO; JPO;			
			DERWENT			

	Туре	L#	Hits	Search Text	DBs	Time Stamp	Comments
1	BRS	L1	146587	preheat\$		2005/01/24 13:01	
2	BRS	L2	15944	shell near3 tube	US- PGPUB; USPAT; EPO; JPO; DERWEN	2005/01/24 13:01	
3	BRS	L3	1771	L1 and L2		2005/01/24 13:01	
4	BRS	L6	550	562/598.ccls.		2005/01/24 13:01	
5	BRS	L7	0	L6 and L4		2005/01/24 13:01	
6	BRS	L9	359007		US- PGPUB; USPAT; EPO; JPO; DERWEN T	2005/01/24 13:01	

7	BRS	L10	12311	L1 same L9	US - PGPUB; USPAT; EPO; JPO; DERWEN T	2005/01/24 13:01	
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9	BRS	L13	0	("6651731").URPN.	USPAT	2005/01/24 13:01	
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11	BRS	L30	1133	L9 same L29	US- PGPUB; USPAT; EPO; JPO; DERWEN T	2005/01/24 13:01	
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17	BRS	L20	1	"4341600".PN.		2005/01/24 13:01	
18	BRS	L21	1	"4365081".PN.		2005/01/24 13:01	

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24	BRS	L5	12	"1089353"	US- PGPUB; USPAT; EPO; JPO; DERWEN	2005/01/24 13:01	
25	BRS	L11	4	L6 and L10	US - PGPUB; USPAT; EPO; JPO; DERWEN	2005/01/24 13:01	
26	BRS	L15	2	("3118927").URPN.		2005/01/24 13:01	
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28	BRS	L17	2	"6676808".pn.		2005/01/24 13:01	
29	BRS	L27	3	"3762465".pn.		2005/01/24 13:01	

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	Туре	L#	Hits	Search Text	DBs	Time Stamp	Comments
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31	BRS	L31	87	L1 same L30		2005/01/24 13:01	
32	BRS	L4	472	L1 same L2	US- PGPUB; USPAT; EPO; JPO; DERWEN	2005/01/24 13:01	

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